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## AN IONIZATION GAUGE

### BACKGROUND OF THE INVENTION

Ionization gauges, more specifically Bayard-Alpert (BA) ionization gauges, are  
5 the most common non-magnetic means of measuring very low pressures and have been  
widely used worldwide since being disclosed in U.S. Pat. No. 2,605,431 in 1952.

A typical ionization gauge includes an electron source, an anode, and an ion  
collector electrode. For the BA gauge, the electron source is located radially outside of  
an ionization space (anode volume) defined by the anode. The ion collector electrode is  
10 disposed within the anode volume. Electrons travel from the electron source toward  
and through the anode, and are eventually collected by the anode. However, in their  
travel, the electrons impact molecules and atoms of gas, constituting the atmosphere  
whose pressure is to be measured, and create ions. The ions are attracted to the ion  
collector electrode by the electric field inside the anode. The pressure of the gas within  
15 the atmosphere can be calculated from ion and electron currents by the formula  $P =$   
 $(I/S)$  (Iion/Ielectron), where S is a constant with the units of 1/torr and is characteristic  
of a particular gauge geometry and electrical parameters.

### SUMMARY OF THE INVENTION

20 The operational lifetime of a typical ionization gauge is approximately ten years  
when the gauge is operated in benign environments. However, these same gauges and

electron sources (filaments) fail in minutes or hours when operated at too high a pressure or in gas types that degrade the emission characteristics of the electron source. Examples of such filament interactions leading to decreased operational lifetime range from degradation of the electron emission properties of the oxide coating on the  
5 filament to exposure to water vapor. Degradation of the oxide coating dramatically reduces the number of electrons generated by the filament, and exposure to water vapor results in the complete burnout of a tungsten filament.

Residual gas analyzers (RGA) are ionization gauges that determine the type of gas present and display the partial pressure of each gas type component which can also  
10 be compiled for a calculation of total gas pressure. RGAs measure the mass-to-charge ratio of ions present and convert the ratio to a signal. However, RGAs measure additional signal peaks that are entirely spurious to the gas species being detected. These peaks form a background spectrum that is significant in magnitude at the lower pressure ranges of a high vacuum. This spurious spectrum is caused by interaction of  
15 some atoms and/or molecules with the material of the filament, thus forming other compounds not originally in the gas phase.

There is provided an ionization gauge having at least one electron source for generating electrons, a collector electrode for collecting ions formed by the impact between the electrons and gas molecules, and an electron window which isolates the  
20 electron source(s) from the gas molecules. The ionization gauge can have an anode which defines an anode volume and retains the electrons in a region of the anode. The collector electrode can be located within the anode volume or outside the anode volume. The ionization gauge can have a mass filter for separating the ions based on mass-to-charge ratio. The ionization gauge can be a Bayard-Alpert type that measures pressure  
25 or a residual gas analyzer that also determines a gas type.

An acceleration electrode can be located between the electron source(s) and the electron window to accelerate the electrons to an energy which allows the electrons to be transmitted through the electron window, and a deceleration electrode can be located between the electron window and the collector electrode to decelerate the electrons to

the desired energy distribution to allow ion formation by the impact between the electrons and the gas molecules, providing ions for the collector electrode to collect so a pressure of the gas can be measured or a type of gas can be determined. An anode or a mass filter can be between the deceleration electrode and the collector. The  
5 acceleration electrode is maintained at an electric potential so that the potential difference between the electron source(s) and the acceleration electrode is at a range of 100 volts to 10,000 volts. The electron window is at a potential comparable to the acceleration electrode potential. The deceleration electrode is maintained at an electric potential so that the potential difference between the electron window and the  
10 deceleration electrode is at a range of 0 volt to 10,000 volts. A second collector electrode can be between the electron window and the deceleration electrode for collecting ions formed by the impact between the electrons and the gas molecules for high pressure conditions of very short mean free paths.

A shield can define a shielded volume so potentials external to the shield do not  
15 disturb the electric charge distribution within the shielded volume. The shield is at least partially open to permit transfer of the gas molecules into the shielded volume. The shielded volume houses the electron source(s), the collector electrode(s), and the electron window. The shield is maintained at a reference potential, where the reference potential can be ground potential.

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#### BRIEF DESCRIPTION OF THE DRAWINGS

The foregoing and other objects, features and advantages of the invention will be apparent from the following more particular description of preferred embodiments of the invention, as illustrated in the accompanying drawings in which like reference  
25 characters refer to the same parts throughout the different views. The drawings are not necessarily to scale, emphasis instead being placed upon illustrating the principles of the invention.

Fig. 1 is a schematic view of a generalized ionization gauge of the present invention.

Fig. 2 is a detailed schematic view of a non-nude type ionization gauge of Fig. 1.

Fig. 3 is a schematic view of a generalized mass analyzer of the present invention.

## 5 DETAILED DESCRIPTION OF THE INVENTION

A description of preferred embodiments of the invention follows.

Generally, as shown in FIG. 1, one ionization gauge 100 of the present invention has an isolation chamber 110 and a measurement chamber 120. The isolation chamber contains at least one electron source 140 and at least one acceleration electrode 150.

10 The measurement chamber 120 contains at least one deceleration electrode 170, an anode 180, and a collector electrode 190. The two chambers 110, 120 are separated by an isolation material 130 which prevents molecules and atoms of gas within the measurement chamber 120 from entering the isolation chamber 110 and degrading the electron source(s) 140. The isolation material 130 has an electron window 160 which  
15 allows electrons to be transmitted from the isolation chamber 110 into the measurement chamber 120. Although the ionization gauge 100 is shown with an anode 180 and collector electrode 190 these components are not needed in all embodiments of the invention as described below. In one embodiment, the ionization gauge 100 is a Bayard-Alpert type gauge.

20 FIG. 2 shows a specific non-nude type ionization gauge 200 embodying the present invention. Although a non-nude type gauge is shown, it should be understood that a nude type ionization gauge can be employed using the principals of the present invention. The ionization gauge 200 has similar components to the ionization gauge 100 (FIG. 1) described above with the following additions. The ionization gauge is  
25 housed in a tube 205 open at one end 225 to allow molecules and atoms of gas to enter the measurement chamber 120 through a shield 220. The shield 220 and tube 205 form a shield volume. An optional second ion collector 210 is added for high pressure measurements of very short mean free paths.

In operation, molecules and atoms of gas enter the measurement chamber 120 through the partially open shield 220. The shield 220 prevents potentials external to the shield 220 from disturbing the electric charge distribution within the measurement chamber 120. The shield 220 is maintained at a reference potential. In one  
5 embodiment, the reference potential is ground potential.

The electron source(s) (for example, filaments) 140 generates electrons (represented by an electron beam 208) within the isolation chamber 110. The electrons are used in ionizing the gas molecules in the measurement chamber 120. The geometrical shape of the filament 140 can be a linear ribbon, a linear wire, a straight  
10 ribbon, a curved ribbon, a straight wire, a hairpin wire, or any other shape known in the art. In one embodiment, the filament 140 is resistively heated to incandescence with an electrical current from an emission regulator (not shown). The thermionically, or otherwise, emitted electrons are accelerated by acceleration electrodes 150 to an energy which allows the electrons to be transmitted through the electron window 160. The  
15 acceleration electrodes 150 operate at an electric potential so that the potential difference between the electron source and the acceleration electrodes is within a range from 100 volts to 10,000 volts depending upon the size, thickness and type of material used for the electron window. In one embodiment, the acceleration electrodes 150 are maintained at a potential difference from the electron source(s) around 1000 volts. The  
20 electron window can be made from aluminum such as described in J.R. Young, "Penetration of Electrons and Ions," *Journal of Applied Physics*, January 1956, the entire teachings of which are herein incorporated by reference.

The electrons must have their energy controlled to a desired level to allow for ionization to occur after the electrons have been transmitted through the electron  
25 window 160. Ionization occurs over an energy spread both higher and lower than the nominal design energy; see Section 5.7 on ionization gauges in *Scientific Foundations of Vacuum Technique* by Saul Dushman, 1962. Ion formation typically occurs at an electron energy around 150 electron volts for nitrogen. Therefore, the energy of the electrons is modified by deceleration electrodes 170 to allow for ionization. The

deceleration electrodes 170 operate at an electric potential so that the potential difference between the electron window and the deceleration electrodes are within a range from 0 volt to 10,000 volts depending upon the incident electron energy required by the size, thickness, and type of window and also the type of pressure measurement.

5 For example, as shown in FIG. 2, the deceleration electrodes 170 are maintained at a potential which reduces the energy of the electrons to 150 electron volts.

The anode grid 180 defines an anode volume 185 and is charged to a positive 180 volts with reference to ground. The anode grid 180 can be made of a wire mesh or like design to allow the electrons to enter and exit the anode grid 180. Most electrons  
10 do not strike the anode grid 180 immediately but pass through the anode grid 180 and into the anode volume 185 where they create ions through electron impact ionization. The electrons which do not strike the anode grid 180 or ionize any molecules, pass through the anode volume 185 into the region between the anode grid 180 and the tube 205. There the electrons decelerate and re-accelerate back toward the grid 180 because  
15 of the electric field created between the tube 205 and the anode grid 180. The electrons continue circulating in this fashion until the electrons are collected by the anode grid 180 or lost to other surfaces. The multiple passes of electrons increase the ionization efficiency of the electron current relative to single pass configurations.

The ions, once created by electron impact ionization, tend to stay within the  
20 anode grid 180. The ions formed within the anode volume 185 are directed by the electric field produced by the potential difference between (a) the anode grid 180 at a potential that is positive with respect to ground and (b) the collector electrode 190 which is at a potential which is near ground potential (i.e., negative relative to the anode potential). This electric field directs the ions to the collector where they are collected to  
25 provide the ion current used to determine the pressure of the gas. In some embodiments, higher pressures can be measured using the second ion collector 210 which is very close to the window for capture of ions formed very close to the window, i.e., not inside the anode, due to short mean free paths at higher pressures..

FIG. 3 shows a residual gas analyzer (RGA) 300 embodying to the present invention. The RGA 300 and ionization gauges 100, 200 described above are essentially the same except the anode grid 180 and collector electrode 190 of FIGS. 1 and 2 are replaced by a mass filter 310 and an ion detector 320.

5           In operation, molecules and atoms of gas enter the measurement chamber 120. The electron source(s) (for example, filaments) 140 generates electrons within the isolation chamber 110. The thermionically emitted electrons (or electrons produced via field emission, photoemission, plasma extraction, from a radioactive source, or otherwise) are accelerated by acceleration electrodes 150 to an energy which allows the  
10   electrons to be transmitted through the electron window 160. The electrons are used in ionizing the gas molecules and atoms in the measurement chamber 120.

          The electrons must have their energy controlled to an appropriate level to allow for ionization to occur after the electrons have been transmitted through the electron window 160. The energy of the electrons is modified by deceleration electrodes 170 to  
15   allow for ionization.

          The ions enter a mass filter 310 which separates the ions based on their mass-to-charge ratio ( $m/z$ ). The mass filter 310 allows selected ions to move to the ion detector 320. The mass filter 310 filters the ions based on their mass-to-charge ratio. Only the ions that match the mass filter parameters are passed through the mass filter 310 at a  
20   given time. The ion detector 310 counts the ions and a signal is generated that is proportional to the total number of ions at each given time as noted just above. The signal is reported to a data system (not shown) which outputs the type of gas present.

          While this invention has been particularly shown and described with references to preferred embodiments thereof, it will be understood by those skilled in the art that  
25   various changes in form and details may be made therein without departing from the scope of the invention encompassed by the appended claims.